

March 10, 2017

BY RESS & COURIER

Ms. Kirsten Walli Board Secretary Ontario Energy Board Suite 2700, 2300 Yonge Street Toronto, Ontario M4P 1E4

Dear Ms. Walli:

Re: Union Gas Limited ("Union") 2017 Storage Enhancement Project Board File #EB-2016-0322

As request by Board Staff at page 8 of their submissions dated February 21, 2017 for the 2017 Storage Enhancement Project, please find attached a copy of a letter that Stantec Consulting send to Mr. and Mrs. Smit regarding the water well monitoring program that has now been completed on their property.

The water well monitoring program determined that the Ontario Drinking Water standards were not exceeded for any health related parameters. The testing also determined that the methane was from a naturally occurring biogenic source not a thermogenic source. Stantec also provided Mr. and Mrs. Smit with some recommendations in regard to venting their water well.

If you have any questions nor required additional information please contact

Sincerely,

W.T. (Bill) Wachsmuth, RPF Senior Administrator, Regulatory Projects :sb

cc: Nancy Marconi, OEB Zora Crnojacki, OEB Andrew Mandyam, Enbridge Tania Persad, Enbridge Erin Henderson, Hydro One Demetrius Kappos, MNRF



Stantec Consulting Ltd. 100-300 Hagey Boulevard, Waterloo ON N2L 0A4

March 8, 2017 File: 1609-00864

Attention: Mr. and Mrs. Smit

Dear Jake and Mary Smit,

Reference: Residential/Private Well Monitoring Program – Union Gas

On behalf of Union Gas Ltd., Stantec Consulting Ltd. (Stantec) is pleased to provide you with the analytical results of the water sample collected from your residential well (RW5) at 3232 Edys Mills Line on January 18, 2017. The water quality sample from your residence was collected from your kitchen tap with an additional water sample for isotopic analyses collected from your garage tap since the hose fitting required to collect the sample did not fit onto your kitchen tap. Stantec understands that both the garage tap and the kitchen tap deliver water that has not been treated. Based on the well questionnaire, Stantec also understands that the water from your well is not used for drinking water but is used for other household purposes.

Table 1 provides a summary of your current and historical water quality results for water sampled from your well. The 2017 water quality results are consistent with concentrations from your well measured in November 2007 and do not suggest changing conditions.

Table 1 provides a comparison of the data to the Ontario Drinking Water Standards (ODWS). The water quality results indicate that the ODWS Maximum Acceptable Concentration (MAC) was not exceeded for the tested health-related parameters.

Although sodium does not have a ODWS MAC, elevated sodium may be a concern for consumers on a sodium restricted diet and as such, the Medical Officer of Health (MOH) set a reporting limit of 20 mg/L. If you decide to use this water as your drinking water source, please advise your doctor of the water quality results at your next visit.

The water quality results from January 18, 2017 also indicate that chloride, iron, turbidity, total dissolved solids (TDS), hardness, and methane exceeded their respective ODWS:

- Chloride, turbidity, TDS, and iron are ODWS Aesthetic Objectives (AO) and are not considered a health-related risk under the ODWS.
- Hardness exceeded its respective ODWS Operational Guideline (OG). This parameter is not considered a health-related risk under the ODWS.
- Methane exceeded the ODWS AO and is discussed in further detail below.

Design with community in mind



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Reference: Residential/Private Well Monitoring Program – Union Gas

METHANE GAS CONCENTRATION

The ODWS for methane is not a health-related parameter but is an aesthetic concern. The ODWS indicate that the aesthetic objective is based on gas bubble release and potential violent spurting from taps for methane concentrations above 3.0 L/m³ (about 60 ppm_v). Methane bubbles in the sample tubing as well as spurting from the kitchen tap were observed during sample collection.

The ODWS indicate that if methane is allowed to accumulate in confined areas, the potential for explosive combustion exists. At concentrations of 50,000 ppm_v to 150,000 ppm_v, methane can accumulate and become an explosive concern under some conditions. At levels below 50,000 ppm_v, which is the lower explosive limit (LEL), the concentration is too dilute for an explosion to occur.

In order to assess the methane risk, the Ministry of the Environment and Climate Change (MOECC) has developed criteria for methane vapour concentration. The MOECC (2016¹) guideline for landfill gas considers that concentrations of methane greater than 10% of the LEL (5,000 ppm_v) may be considered hazardous conditions.

Table 1 shows that the aqueous methane concentrations from the sample collected from your well was 45 L/m³, which is the same concentration measured in a sample collected at your well by Stantec on November 26, 2007. Using Henry's Law and the Ideal Gas Law, at 10°C and 100 kPa, the aqueous concentrations would be in equilibrium with a vapour concentration of 890 ppm_v. This calculated vapour concentration does not exceed the safety criteria level (5,000 ppm_v) presented by the MOECC (2016¹) and is <2% of the LEL.

To evaluate the potential source of the methane, a sample was submitted for isotopic analyses of the carbon and hydrogen (²H and ¹³C) in the dissolved methane. The isotopic composition of methane is a diagnostic tool to aid in characterization of the origin of natural gas. Depending on the formation process of natural gas, the isotopic signature of the hydrogen and carbon in the methane will vary. Methane can be produced by:

- a low temperature, bacteria driven process (biogenic) of decomposition of organic matter (Schoell, 1980²)
- a higher temperature, high pressure geological driven process (thermogenic).

The natural gas (methane) in Union Gas's storage pools is sourced from Alberta and Pennsylvania and is of thermogenic origin.

² Schoell, M., 1980. Genetic Characterization of Natural Gases. American Association of Petroleum Geologists, Bulletin v 67.

¹ MOECC, 2016. D-4-1 Assessing Methane Hazards from Landfill Sites.



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Reference: Residential/Private Well Monitoring Program – Union Gas

The isotopic signature of the ¹³C within methane generally ranges from -50‰ to -20‰ for thermogenic methane and -110‰ to -50‰ for biogenic methane (Whiticar, 1999³). The result from the sample collected from your well was -67.7‰, which falls within the expected range for biogenic methane.

The isotope signature of the ²H within methane generally ranges from approximately -290‰ to -100‰ for thermogenic methane and -400‰ to -280‰ for biogenic methane (Whiticar, 1999³). The result from the sample at collected from your well was -299‰, with a lab repeat sample result of -290‰, which falls within the expected range for biogenic methane and just outside of the range for thermogenic methane.

Another useful distinction between thermogenic and biogenic methane is the proportions of other larger hydrocarbons present within the gas such as propane and ethane. The dissolved gas within the groundwater sample at your well was primarily methane with a trace of ethane. Propane was not detected above the laboratory quantification limit. This ratio of dissolved gas concentrations is also indicative of biogenic methane (Sharma, 2013⁴).

Based on the methane analysis, Stantec concludes that the methane from your groundwater sample is representative of naturally occurring biogenic methane, and not from a thermogenic source.

The water quality data presented in this letter and in the attached table are representative of water quality in your well on the day the data were collected. Water quality in your well may change over time. Stantec recommends that you follow the Public Health Ontario guidelines for sampling and caring for your well. Information regarding well water quality, maintenance and disinfection can be found at the Public Health Ontario web page

(http://www.publichealthontario.ca/en/ServicesAndTools/LaboratoryServices/Pages/Watertesting.aspx), or contact the Lambton Public Health Unit at 1-519-383-8331 or Public Health Ontario at 1-877-543-8931.

As detailed in the Stantec letter dated January 26, 2017, Stantec recommends that you retain a licensed well contractor to bring your well in compliance with O.Reg.903 regarding well venting.

³ Whiticar, M.J., 1999. Carbon and Hydrogen Isotope Systematics of Bacterial Formation and Oxidation of Methane. Chemical Geology, 161 (1999) 291 to 314.

⁴ Sharma, S., Mulder, M.L., Sack, A., Schroeder, K. and Hammack, R. Isotope Approach to Assess Hydrologic Connections During Marcellus Shale Drilling. Groundwater (2013) 1 to 10.



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Reference: Residential/Private Well Monitoring Program – Union Gas

We thank you for your participation in the Union Gas well monitoring program.

Regards,

STANTEC CONSULTING LTD.

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Attachment: Table 1 - Groundwater Analytical Results RW5

c. Evan Tomek, Doug Schmidt - Union Gas

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Reference: Residential/Private Well Monitoring Program – Union Gas

ATTACHMENT: TABLE 1 – GROUNDWATER ANALYTICAL RESULTS RW5

Table 1 Summary of Groundwater Analytical Results - RW5 Storage Pool Enhancement Project Union Gas

Sample Location				RW5	I.
Sample Date			26-Nov-07	18-Jan-17	18-Jan-17
Sample ID			WG14090044220071126-RW5	WG-160900864-20170118-JK1	WG-160900864-20170118-JK
					Lab-Dup
ampling Company			STANTEC	STANTEC	STANTEC
aboratory			MAXX	MAXX / ITT	MAXX
aboratory Work Order			A7D2337	B711312 / 170019	B711312
Laboratory Sample ID			W11949	DTV059 / 42894	DTV059
Sample Type	Units	ODWS			Lab Replicate
lealth Related parameters					
Antimony	mg/L	0.006 ^B	<0.0005	<0.00050	-
Arsenic	mg/L	0.025 ^B	< 0.005	<0.0010	-
Barium	mg/L	1 ^C	0.66	0.69	_
Boron		5 ⁸	2.3	2.3	
	mg/L	0.005 ^C			-
	mg/L	0.05 ^C	<0.0001	<0.00010	-
Chromium	mg/L		<0.005	<0.0050	-
ead	mg/L	0.01 _c ^C	<0.0005	<0.00050	-
litrate (as N)	mg/L	10.0 _d ^C	<0.1	<0.10	-
vitrite (as N)	mg/L	1.0 _d ^C	<0.01	<0.010	-
elenium	mg/L	0.01 ^C	< 0.002	<0.0020	-
Iranium	mg/L	0.02 ^C	< 0.0001	<0.00010	_
Aicrobiological Parameters	0.				
_	-1.1000	0 ^A		<u>^</u>	
ischerichia coli (E.Coli)	cfu/100mL		-	0	-
otal Coliform Background	cfu/100mL	n/v	-	32	-
otal Coliforms	cfu/100mL	0 ^A	-	0	-
Aesthetic Parameters					
Chloride	mg/L	250 ^D	740 ^D	790 ^D	-
Copper	mg/L	1 ^D	<0.001	0.0010	-
		5 ^D	0.8	0.88	_
Dissolved Organic Carbon (DOC)	mg/L	0.3 ^D			-
ron	mg/L	0.05 ^D	0.49 ^D	0.50 ^D	-
Manganese	mg/L		0.02	0.028	-
Sodium	mg/L	200g ^D 20g ^F	590 ^{DF}	580 ^{DF}	-
Sulfate	mg/L	500 _h D	<1	<1.0	-
otal Dissolved Solids (Calculated)	mg/L	500 ^D	1, 540^D	1,600 ^D	-
linc	mg/L	5 ^D	<0.005	0.0067	-
Operational Parameters					
Alkalinity, Total (as CaCO3)	mg/L	30-500 ^E	268	270	
		0.1 ^E			-
Aluminum	mg/L	80-100 ^E	<0.005	0.0068	-
Hardness (as CaCO3)	mg/L		98	110 ^E	-
ъН	S.U.	6.5-8.5 ^E	8.1	8.10	-
Sulfide	mg/L	n/v	-	0.32	-
Fotal Suspended Solids	mg/L	n/v	-	<10	<10
Turbidity, Lab	NTU	5, D E	-	6.8 ^D	-
Methane					
Methane (Calculated)	ppmv	10,000 ^G	891	891	-
Vethane	L/m3	3 ^D	45 ^D	45 ^D	
Dissolved Gases	2,1110		40	40	
			1	1	1
Acetylene	L/m3	n/v	-	<0.002	-
thane	L/m3	n/v	-	0.003	-
Ethylene (Ethene)	L/m3	n/v	-	<0.002	-
Propane	L/m3	n/v	-	<0.002	-
Propene	L/m3	n/v	-	<0.002	-
sotope Analysis					
13C Methone (CH4)	none	n/v	_	677/682	-
313C Methane (CH4)	none	n/v		-67.7 / -68.2 -299 / -289	
32H Methane (CH4) Other Parameters	‰	n/v	-	-277 / -207	-
				1	1
Alkalinity, Bicarbonate (as CaCO3)	mg/L	n/v	264	260	-
Alkalinity, Carbonate (as CaCO3)	mg/L	n/v	3	3.1	-
Ammonia (as N)	mg/L	n/v	0.50	0.51	-
Anion Sum	meq/L	n/v	26.2	27.5	-
Beryllium	mg/L	n/v	<0.0005	<0.00050	-
Calcium	mg/L	n/v	25	29	_
Cation Sum	meg/L	n/v	27.8	27.8	_
Cobalt					
	mg/L	n/v	<0.0005	<0.00050	-
Electrical Conductivity, Lab	µmhos/cm	n/v	2,900	3,100	-
on Balance	%	n/v	2.98	0.490	-
angelier Index (at 20 C)	none	n/v	0.331	0.368	-
angelier Index (at 4 C)	none	n/v	0.0860	0.124	-
<i>l</i> agnesium	mg/L	n/v	8.5	9.8	-
- Nolybdenum	mg/L	n/v	0.002	0.0014	-
lickel	mg/L	n/v	<0.001	<0.0010	_
Drthophosphate(as P)	mg/L	n/v	<0.01	<0.010	_
					-
Phosphorus	mg/L	n/v	<0.1	<0.10	-
Potassium	mg/L	n/v	3.6	3.5	-
aturation pH (at 20 C)	none	n/v	7.78	7.73	-
Saturation pH (at 4 C)	none	n/v	8.03	7.97	-
Silicon	mg/L	n/v	4.8	4.1	-
ilver	mg/L	n/v	<0.0001	<0.00010	-
trontium	mg/L	n/v	0.74	0.81	-
Thallium	mg/L	n/v	< 0.00005	<0.000050	-

Titanium	mg/L	n/v	<0.005	<0.0050	-
Vanadium	mg/L	n/v	<0.005	<0.0010 MI	-

Notes:

ODWS	Technical Support Document for Ontario Drinking Water Standards, Objectives and Guidelines (MOE, 2006, revised January 2017)
A	ODWS Table 1 - Microbiological Standards, Maximum Acceptable Concentration
в	ODWS Table 2 - Chemical Standards, Interim Maximum Acceptable Concentration
С	ODWS Table 2 - Chemical Standards, Maximum Acceptable Concentration
D	ODWS Table 4 - Chemical/Physical Objectives and Guidelines, Aesthetic Objectives
E	ODWS Table 4 - Chemical/Physical Objectives and Guidelines, Operational Guidelines
F	ODWS Table 4 - Medical Officer of Health Reporting Limit
G	The LEL for methane is 50,000 ppmv. MOE (2016) presents criteria of 10,000 ppmv
6.5 ^A	Concentration exceeds the indicated standard.
15.2	Measured concentration did not exceed the indicated standard.
<0.50	Laboratory reporting limit was greater than the applicable standard.
< 0.03	Analyte was not detected at a concentration greater than the laboratory reporting limit.
n/v	No standard/guideline value.
-	Parameter not analyzed / not available.
с	This standard applies to water at the point of consumption. Since lead is a component in some plumbing systems, first flush water may contain
	higher concentrations of lead than water that has been flushed for five minutes.
d	Where both nitrate and nitrite are present, the total of the two should not exceed 10 mg/L (as nitrogen).
DF g	The aesthetic objective for sodium in drinking water is 200 mg/L. The local Medical Officer of Health should be notified when the sodium concentration
	exceeds 20 mg/L so that this information may be communicated to local physicians for their use with patients on sodium restricted diets.
h	When sulfate levels exceed 500 mg/L, water may have a laxative effect on some people.
i	Applicable for all waters at the point of consumption.
i	The operational guidelines for filtration processes are provided as performance criteria in the Procedure for Disinfection of Drinking Water in Ontario.
MI	Detection limit was raised due to matrix interferences.

